

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

***N*,4-Dimethyl-*N*-phenylbenzenesulfonamide**

Bin Zhou\* and Peng-Wu Zheng

School of Pharmacy, Jiangxi Science and Technology Normal University, Nanchang 330013, People's Republic of China

Correspondence e-mail: tju\_zhoubin@yahoo.com.cn

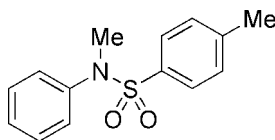
Received 20 October 2007; accepted 10 November 2007

Key indicators: single-crystal X-ray study;  $T = 294$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.043;  $wR$  factor = 0.121; data-to-parameter ratio = 14.2.

The two rings in the title compound,  $\text{C}_{14}\text{H}_{15}\text{NO}_2\text{S}$ , are inclined to each other by an angle of  $43.78(13)^\circ$ . In the crystal structure, molecules are linked by a single weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bond involving an  $\text{S}=\text{O}$  group as acceptor.

## Related literature

For related *N*-arylbenzenesulfonamides, see: Shi (2007); Chang *et al.* (2007); Yu *et al.* (2007); Xing *et al.* (2006); Yu (2006); Xing & Zeng (2005).



## Experimental

## Crystal data

$\text{C}_{14}\text{H}_{15}\text{NO}_2\text{S}$   
 $M_r = 261.33$   
 Monoclinic,  $P2_1/n$   
 $a = 14.156(4)$  Å  
 $b = 6.2251(17)$  Å

$c = 16.295(5)$  Å  
 $\beta = 112.821(5)^\circ$   
 $V = 1323.5(6)$  Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation

$\mu = 0.24$  mm<sup>-1</sup>  
 $T = 294(2)$  K

0.22 × 0.20 × 0.14 mm

## Data collection

Bruker SMART 1K CCD area-detector diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 1997)  
 $T_{\min} = 0.950$ ,  $T_{\max} = 0.968$

6561 measured reflections  
 2341 independent reflections  
 1738 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.031$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.121$   
 $S = 1.04$   
 2341 reflections  
 165 parameters

6 restraints  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.30$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C11}-\text{H11}\cdots\text{O2}^i$	0.93	2.57	3.384 (4)	147

Symmetry code: (i)  $x - \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2144).

## References

- Bruker (1997). SMART (Version 5.611), SAINT (Version 6.0), SADABS (Version 2.03) and SHELXTL (Version 5.10). Bruker AXS Inc., Madison, Wisconsin, USA.
- Chang, J., Zhang, T.-J., Liu, C.-X. & Zhao, X.-M. (2007). *Acta Cryst.* **E63**, o3670.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Shi, F.-E. (2007). *Acta Cryst.* **E63**, o4004.
- Xing, J.-D., Bai, G.-Y., Zeng, T. & Li, J.-S. (2006). *Acta Cryst.* **E62**, o79–o80.
- Xing, J.-D. & Zeng, T. (2005). *Acta Cryst.* **E61**, o4318–o4319.
- Yu, Y.-Y. (2006). *Acta Cryst.* **E62**, o2308–o2309.
- Yu, H.-J., Chen, J.-Z., Simpson, J., Li, J.-S. & Bai, G.-Y. (2007). *Acta Cryst.* **E63**, o3720.

**supplementary materials**

*Acta Cryst.* (2007). E63, o4727 [ doi:10.1107/S1600536807057650 ]

## *N*,4-Dimethyl-*N*-phenylbenzenesulfonamide

B. Zhou and P.-W. Zheng

### Comment

The N atom of the title sulfonamide, (I), has a pyramidal geometry (Fig. 1), but the pyramid is somewhat shallower than expected for pure  $sp^3$  hybridization [C9—N1—C8: 116.86 (18)°; C9—N1—S1: 117.00 (14)°; C8—N1—S1: 116.69 (15)°]. The benzene and phenyl rings are inclined to each other by an angle of 43.78 (13)°. A weak C—H···O hydrogen bond (Table 1) involving a S=O group of sulfonamide links molecules to form chains in the crystal structure. No significant  $\pi$ – $\pi$  interactions are observed in the packing structure.

### Experimental

The title compound was obtained by reaction of *N*-methylaniline with *p*-tosyl chloride in the presence of aqueous sodium bicarbonate. Colourless blocks of (I) were grown by slow evaporation of an ethyl acetate solution at 298 K.

### Refinement

Methyl C atom of the *p*-tolyl group was restrained with a standard deviation of 0.01 Å<sup>2</sup> so that  $U_{ij}$  components approximate to isotropic behaviour. All H atoms were positioned geometrically (C—H = 0.93 Å for aromatic CH, 0.96 Å for methyl CH<sub>3</sub>) and refined as riding with  $U_{iso}(H) = 1.2U_{eq}(C)$  for aromatic CH and  $U_{iso}(H) = 1.2U_{eq}(C)$  for methyl CH<sub>3</sub>.

### Figures

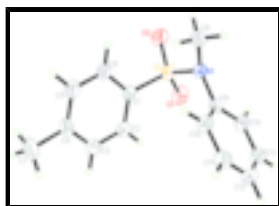


Fig. 1. The molecular structure of (I), with the atom-numbering scheme and 30% probability displacement ellipsoids for non-H atoms.

## *N*,4-Dimethyl-*N*-phenylbenzenesulfonamide

### Crystal data

C<sub>14</sub>H<sub>15</sub>NO<sub>2</sub>S

$M_r = 261.33$

Monoclinic,  $P2_1/n$

Hall symbol: -P 2yn

$a = 14.156$  (4) Å

$F_{000} = 552$

$D_x = 1.312$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 2238 reflections

$\theta = 2.7$ – $25.0^\circ$

# supplementary materials

---

$b = 6.2251 (17) \text{ \AA}$   
 $c = 16.295 (5) \text{ \AA}$   
 $\beta = 112.821 (5)^\circ$   
 $V = 1323.5 (6) \text{ \AA}^3$   
 $Z = 4$

$\mu = 0.24 \text{ mm}^{-1}$   
 $T = 294 (2) \text{ K}$   
Block, colourless  
 $0.22 \times 0.20 \times 0.14 \text{ mm}$

## Data collection

Bruker SMART 1K CCD area-detector diffractometer  
Radiation source: fine-focus sealed tube  
Monochromator: graphite  
 $T = 294(2) \text{ K}$   
 $\varphi$  and  $\omega$  scans  
Absorption correction: multi-scan (SADABS; Bruker, 1997)  
 $T_{\min} = 0.950$ ,  $T_{\max} = 0.968$   
6561 measured reflections

2341 independent reflections  
1738 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.031$   
 $\theta_{\max} = 25.0^\circ$   
 $\theta_{\min} = 1.6^\circ$   
 $h = -9 \rightarrow 16$   
 $k = -7 \rightarrow 7$   
 $l = -19 \rightarrow 14$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.121$   
 $S = 1.04$   
2341 reflections  
165 parameters  
6 restraints  
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0591P)^2 + 0.4321P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.30 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.21 \text{ e \AA}^{-3}$   
Extinction correction: SHELXL97,  
 $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$   
Extinction coefficient: 0.046 (3)

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.86815 (5)	0.54101 (10)	0.07849 (4)	0.0497 (2)
O1	0.83906 (15)	0.7595 (3)	0.06292 (12)	0.0671 (5)
O2	0.96603 (13)	0.4862 (3)	0.14363 (11)	0.0689 (6)
N1	0.86374 (13)	0.4405 (3)	-0.01530 (12)	0.0420 (5)
C1	0.77411 (17)	0.4082 (4)	0.10434 (14)	0.0454 (6)
C2	0.6781 (2)	0.4967 (4)	0.08104 (17)	0.0595 (7)
H2	0.6641	0.6325	0.0554	0.071*
C3	0.6033 (2)	0.3831 (5)	0.09588 (19)	0.0698 (8)
H3	0.5390	0.4449	0.0812	0.084*
C4	0.6209 (2)	0.1798 (5)	0.13204 (17)	0.0623 (7)

C5	0.7185 (2)	0.0976 (4)	0.15728 (18)	0.0623 (7)
H5	0.7331	-0.0366	0.1843	0.075*
C6	0.79442 (19)	0.2084 (4)	0.14360 (16)	0.0533 (6)
H6	0.8596	0.1491	0.1607	0.064*
C7	0.5362 (3)	0.0523 (6)	0.1427 (2)	0.0971 (11)
H7A	0.5218	0.1096	0.1912	0.146*
H7B	0.5570	-0.0951	0.1547	0.146*
H7C	0.4758	0.0608	0.0888	0.146*
C8	0.91396 (19)	0.2314 (4)	-0.01064 (18)	0.0573 (7)
H8A	0.8734	0.1205	0.0006	0.086*
H8B	0.9807	0.2338	0.0366	0.086*
H8C	0.9206	0.2035	-0.0661	0.086*
C9	0.77237 (16)	0.4813 (3)	-0.09328 (14)	0.0400 (5)
C10	0.69764 (18)	0.3281 (4)	-0.12578 (16)	0.0547 (7)
H10	0.7039	0.1977	-0.0963	0.066*
C11	0.6130 (2)	0.3688 (5)	-0.20252 (18)	0.0672 (8)
H11	0.5627	0.2640	-0.2253	0.081*
C12	0.6022 (2)	0.5600 (5)	-0.24519 (17)	0.0650 (7)
H12	0.5443	0.5871	-0.2963	0.078*
C13	0.6769 (2)	0.7121 (5)	-0.21258 (18)	0.0637 (7)
H13	0.6697	0.8432	-0.2418	0.076*
C14	0.76271 (19)	0.6736 (4)	-0.13699 (16)	0.0515 (6)
H14	0.8139	0.7771	-0.1156	0.062*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0567 (4)	0.0452 (4)	0.0430 (4)	-0.0123 (3)	0.0148 (3)	-0.0036 (3)
O1	0.1001 (14)	0.0396 (9)	0.0658 (12)	-0.0125 (9)	0.0367 (10)	-0.0088 (8)
O2	0.0501 (10)	0.0910 (14)	0.0489 (11)	-0.0176 (9)	0.0008 (8)	0.0022 (9)
N1	0.0419 (10)	0.0403 (10)	0.0425 (11)	-0.0045 (8)	0.0149 (8)	-0.0001 (8)
C1	0.0520 (14)	0.0445 (13)	0.0380 (12)	-0.0004 (10)	0.0155 (10)	-0.0003 (10)
C2	0.0616 (17)	0.0583 (16)	0.0557 (16)	0.0072 (12)	0.0195 (13)	0.0131 (12)
C3	0.0538 (17)	0.095 (2)	0.0619 (18)	0.0071 (15)	0.0239 (14)	0.0123 (16)
C4	0.0694 (18)	0.0772 (19)	0.0490 (15)	-0.0138 (15)	0.0325 (13)	-0.0015 (13)
C5	0.082 (2)	0.0521 (15)	0.0633 (17)	0.0013 (14)	0.0399 (15)	0.0081 (13)
C6	0.0588 (15)	0.0502 (14)	0.0550 (15)	0.0042 (11)	0.0266 (12)	0.0057 (11)
C7	0.090 (2)	0.122 (3)	0.098 (3)	-0.027 (2)	0.058 (2)	0.005 (2)
C8	0.0600 (16)	0.0511 (15)	0.0643 (17)	0.0055 (12)	0.0280 (13)	0.0010 (12)
C9	0.0427 (12)	0.0415 (12)	0.0385 (12)	-0.0046 (9)	0.0186 (10)	-0.0017 (9)
C10	0.0569 (15)	0.0532 (15)	0.0498 (15)	-0.0142 (12)	0.0162 (12)	0.0030 (12)
C11	0.0539 (16)	0.083 (2)	0.0584 (17)	-0.0234 (14)	0.0150 (13)	-0.0073 (15)
C12	0.0531 (16)	0.090 (2)	0.0455 (15)	0.0020 (15)	0.0118 (12)	0.0092 (15)
C13	0.0734 (18)	0.0612 (17)	0.0567 (17)	0.0089 (14)	0.0253 (14)	0.0161 (13)
C14	0.0584 (15)	0.0447 (14)	0.0511 (15)	-0.0059 (11)	0.0210 (12)	0.0029 (11)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

S1—O1	1.4152 (18)	C7—H7A	0.9600
-------	-------------	--------	--------

## supplementary materials

---

S1—O2	1.4219 (18)	C7—H7B	0.9600
S1—N1	1.6300 (19)	C7—H7C	0.9600
S1—C1	1.751 (2)	C8—H8A	0.9600
N1—C9	1.441 (3)	C8—H8B	0.9600
N1—C8	1.471 (3)	C8—H8C	0.9600
C1—C2	1.376 (3)	C9—C10	1.369 (3)
C1—C6	1.377 (3)	C9—C14	1.372 (3)
C2—C3	1.370 (4)	C10—C11	1.379 (3)
C2—H2	0.9300	C10—H10	0.9300
C3—C4	1.378 (4)	C11—C12	1.357 (4)
C3—H3	0.9300	C11—H11	0.9300
C4—C5	1.378 (4)	C12—C13	1.365 (4)
C4—C7	1.503 (4)	C12—H12	0.9300
C5—C6	1.365 (4)	C13—C14	1.375 (3)
C5—H5	0.9300	C13—H13	0.9300
C6—H6	0.9300	C14—H14	0.9300
O1—S1—O2	119.73 (11)	H7A—C7—H7B	109.5
O1—S1—N1	107.34 (10)	C4—C7—H7C	109.5
O2—S1—N1	106.24 (11)	H7A—C7—H7C	109.5
O1—S1—C1	107.52 (12)	H7B—C7—H7C	109.5
O2—S1—C1	108.60 (11)	N1—C8—H8A	109.5
N1—S1—C1	106.75 (10)	N1—C8—H8B	109.5
C9—N1—C8	116.86 (18)	H8A—C8—H8B	109.5
C9—N1—S1	117.00 (14)	N1—C8—H8C	109.5
C8—N1—S1	116.69 (15)	H8A—C8—H8C	109.5
C2—C1—C6	119.7 (2)	H8B—C8—H8C	109.5
C2—C1—S1	120.42 (19)	C10—C9—C14	120.0 (2)
C6—C1—S1	119.78 (18)	C10—C9—N1	120.9 (2)
C3—C2—C1	119.5 (2)	C14—C9—N1	119.07 (19)
C3—C2—H2	120.2	C9—C10—C11	119.5 (2)
C1—C2—H2	120.2	C9—C10—H10	120.2
C2—C3—C4	121.7 (3)	C11—C10—H10	120.2
C2—C3—H3	119.1	C12—C11—C10	120.7 (2)
C4—C3—H3	119.1	C12—C11—H11	119.6
C3—C4—C5	117.5 (2)	C10—C11—H11	119.6
C3—C4—C7	121.0 (3)	C11—C12—C13	119.6 (2)
C5—C4—C7	121.5 (3)	C11—C12—H12	120.2
C6—C5—C4	121.7 (3)	C13—C12—H12	120.2
C6—C5—H5	119.2	C12—C13—C14	120.6 (2)
C4—C5—H5	119.2	C12—C13—H13	119.7
C5—C6—C1	119.7 (2)	C14—C13—H13	119.7
C5—C6—H6	120.1	C9—C14—C13	119.5 (2)
C1—C6—H6	120.1	C9—C14—H14	120.2
C4—C7—H7A	109.5	C13—C14—H14	120.2
C4—C7—H7B	109.5		
O1—S1—N1—C9	-48.66 (18)	C3—C4—C5—C6	-2.9 (4)
O2—S1—N1—C9	-177.86 (15)	C7—C4—C5—C6	176.6 (3)
C1—S1—N1—C9	66.39 (17)	C4—C5—C6—C1	0.6 (4)

O1—S1—N1—C8	165.89 (16)	C2—C1—C6—C5	1.4 (4)
O2—S1—N1—C8	36.69 (19)	S1—C1—C6—C5	-174.82 (19)
C1—S1—N1—C8	-79.07 (18)	C8—N1—C9—C10	44.9 (3)
O1—S1—C1—C2	19.8 (2)	S1—N1—C9—C10	-100.5 (2)
O2—S1—C1—C2	150.7 (2)	C8—N1—C9—C14	-132.6 (2)
N1—S1—C1—C2	-95.2 (2)	S1—N1—C9—C14	82.0 (2)
O1—S1—C1—C6	-164.01 (19)	C14—C9—C10—C11	0.0 (4)
O2—S1—C1—C6	-33.1 (2)	N1—C9—C10—C11	-177.5 (2)
N1—S1—C1—C6	81.1 (2)	C9—C10—C11—C12	-1.1 (4)
C6—C1—C2—C3	-1.1 (4)	C10—C11—C12—C13	1.1 (4)
S1—C1—C2—C3	175.2 (2)	C11—C12—C13—C14	-0.1 (4)
C1—C2—C3—C4	-1.4 (4)	C10—C9—C14—C13	1.0 (4)
C2—C3—C4—C5	3.3 (4)	N1—C9—C14—C13	178.5 (2)
C2—C3—C4—C7	-176.2 (3)	C12—C13—C14—C9	-1.0 (4)

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C11—H11 $\cdots$ O2 <sup>i</sup>	0.93	2.57	3.384 (4)	147

Symmetry codes: (i)  $x-1/2, -y+1/2, z-1/2$ .

Fig. 1

